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Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

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To cite this Article Mastryukova, Tatyana A. , Aladzheva, Inga M. , Lobanov, Dmitrii I. , Bykhovskaya, Olga V. , Petrovskii, Pavel V. , Lyssenko, Konstantin A. and Kabachnik, Martin I.(1999) 'Synthesis and Some Transformations of 1,2-Heterophosphacyclanes', Phosphorus, Sulfur, and Silicon and the Related Elements, 144: 1, 569-572

To link to this Article: DOI: 10.1080/10426509908546308 URL: http://dx.doi.org/10.1080/10426509908546308

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Synthesis and Some Transformations of 1,2-Heterophosphacyclanes

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The general pathway to 1,2-monoheterophosphacyclanes via intramolecular S-and N-alkylation of ω -halogenoalkylsubstituted thiophosphoryl and iminophosphoryl compounds has been developed. Intramolecular alkylation of 3-and 4-halogenoalkyldiphenylphosphine sulfides results in $1,2\lambda^4$ -thiaphospholanium and thiaphosphorinanium halogenides. In solution of these compounds the rare ring-chain halogenotropic tautomerism has been observed and investigated in detail. Intramolecular Pishchimuka rearrangement of the esters of ω -halogenoalkylsubstituted thiophosphorus acids was found to be a synthetic route to 2-oxo-1,2 λ^5 -thiaphospholanes and thiaphosphorinanes.

Keywords: intramolecular alkylation; 1,2-thiaphosphacyclanes; ring-chain halogenotropic tautomerism; 1,2-azaphosphacyclanes

INTRODUCTION

Although the heterocyclic organophosphorus compounds are known for decades 1,2-monoheterophosphacyclanes with C-P-S and C-P-N fragments are poorly investigated. Contrary to 1,3,2-diheterophophacyclanes, there were no facile synthetic routes for their preparation. We elaborated the general method of 1,2-heterophosphacyclanes synthesis by intramolecular alkylation of ω-halogenoalkylsubstituted compounds with P=S and P=N groups.

RESULTS AND DISCUSSION

Intramolecular Alkylation Of $\,\omega$ - Halogenoalkyldiphenylphosphine Sulfides

Recently^[1] we have found that refluxing of 3- and 4- chloroalkyldiphenylphosphine sulfides 1a,b (throughout this paper a: n=3; b: n=4) with sodium iodide in acetone leads to 2,2-diphenyl-1,2 λ^4 -thiaphospholanium and thiaphosphorinanium iodides 2a,b (Hlg = 1).

The reaction proceeds initially by replacing chlorine atom in 1 by iodine to give 3. The intramolecular alkylation of 3 produces cyclic thiaphosphonium iodides 2 with high yields. We failed to cyclicyze 3-chloropropylphosphine sulfide 1a even under heating without solvent at 140°C during 2 hours. Contrary to iodides 3, ω-bromoalkyldiphenylphosphine sulfides 4a,b are rather stable substances. Nevertheless both crystalline bromide 4a^[2] and liquid bromide 4b turn completely to cyclic bromides 2a,b on heating at 100°C within 0.5 hour. Perchlorates and tetrafluoroborates 6 were prepared by anion exchange. The structures of compounds 1,2,4,6 were determined by IR, Raman, ¹H and ³¹P spectroscopy and confirmed by X-ray crystal analysis. Hydrolysis of the salts 2a,b at 100°C under argon proceeds as a cleavage of the P-S bond yielding phosphinyl substituted mercaptanes Ph₂P(O)(CH₂)_nSH (7a,b); the reaction with sodium ethylate in ethanol-benzene in air leads to disulfides [(Ph₂P(O)(CH₂)_nS]₂ (8a,b).

Ring-Chain Halogenotropic Tautomerism

The behaviour of the cyclic salts 2 and bromides 4 in solution was unexpected. Two tautomeric forms were found to exist in solution: the ring $1.2\lambda^4$ -thiaphospholanium 2a (or thiaphosphorinanium 2b) form and the isomeric open phosphine sulfide form 1,3 or 4.

It is a new type of scantily investigated ring-chain anionotropic tautomerism. The equilibrium for bromides 2 and 4 does not depend on which isomer is taken for the preparation of the solution. The equilibrium position is influenced by the ring size, the solvent used and the anion nature (Table 1). The cyclic form content in equilibrium increases when passing from five-membered 1,2-thiaphosholanium ring to the less strained six-membered 1,2-thiaphosphorinanium one. The equilibrium shifts to the cyclic form in more polar acetonitrile as compared with chloroform and methylene chloride. In passing from iodides to bromides and further to chlorides the content of the open form increases.

TABLE 1. THE CONTENT OF THE CYCLIC FORM 2 AND 6 (%)3)

Solvent	Chlorides		Bromides		Iodides		Perchlorates	
	n=3	n=4	n=3	n=4	n=3	n=4	n=3	n=4
CH ₂ Cl ₂	0	0	38	48	70	85	100	100
CHCl ₃	0	0	65	85	87	92	100	100
CH ₃ CN	9ь	3c	82	82	100	100	-	-

a) ³¹P NMR data; b) after 6 months; c) after 6 months the equilibrium position didn't achieve.

The temperature dependence of equilibrium in methylene chloride was investigated using bromide 4a as an example. An elevation of the temperature was found to result in increasing of the open form. Some thermodynamical parameters of tautomeric equilibrium were calculated. By ³¹P NMR method the kinetics of tautomeric transformations of bromides 2,4 was studied in methylene chloride at 20°C. Both cyclic and open isomers were used as starting substances. In the case of bromides 2a, 4a the equilibrium was reached after about 25 hours while for bromides 2b, 4b - after about 25 days. The rate constants k₁ and k₂ of the direct and reverse reaction were obtained from the experimental data.

Intramolecular Pishchimuka Rearrangement

To elaborate the synthetic routes to new types of 1,2-thiaphosphacyclanes we extended intramolecular alkylation over another classes of organothiophosphorus

compounds. It was found that the esters of thiophsphorus acids 9-11 underwent the intramolecular rearrangement on refluxing with sodium iodide producing 2-oxo-1,2 λ 5-thiaphosphacyclanes (thiolphostones) 12-14.

The idea can be successfully used for synthesis of 1,3,2-dithiolphostones.

$$\begin{array}{c} R \\ \text{EtO} \\ \textbf{9} \text{ a, b - 11 a, b} \\ \\ \hline \\ \text{EtO} \\ \\ \textbf{EtO} \\ \\ \textbf{P(S)(CH}_2)_{\text{nCI}} \\ \hline \\ \text{reflux} \\ \hline \\ \text{reflux} \\ \\ \hline \\ \textbf{EtO} \\ \\ \textbf{P(S)(CH}_2)_{\text{nI}} \\ \\ \hline \\ \textbf{EtO} \\ \\ \textbf{P(S)(CH}_2)_{\text{nI}} \\ \\ \hline \\ \textbf{EtO} \\ \\ \textbf{CH}_2 \\ \hline \\ \textbf{CH}_2 \\ \hline \\ \textbf{CH}_2 \\ \hline \\ \textbf{CH}_2 \\ \hline \\ \textbf{O(CH}_2)_{\text{n-2}} \\ \\ \textbf{12} \text{ a, b - 14 a, b} \\ \\ \textbf{9, 12: R = EtO; 10, 13: R = Ph; 11, 14: R = Et} \\ \end{array}$$

The method of intramolecular alkylation has been advantageously extended also to the synthesis of 1,2-azaphosphocyclanes. The reaction of 3-chloropropyldiphenylphosphine with phenyl azide resulted in phosphine imine 15 which turned into cyclic 1-2 λ^4 -azaphospholanium chloride 16 at 20°C. By refluxing of 16 with sodium iodide in acetone iodide 17 was produced. Its structure was confirmed by X-ray analysis.

Acknowledgements

This work was supported by the Russian Fundamental Research Foundation (grants No 96-03-32992, No 96-15-97298).

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